

## Reading list, Chemistry 8565 – Chemical Reaction Dynamics

### ***Reading for Monday, January 22, 2018***

“Potential Energy Surfaces,”

D. G. Truhlar, in *The Encyclopedia of Physical Science and Technology*, 3rd edition, edited by R. A. Meyers (Academic Press, New York, 2001), Vol. 13, pages 9-17.  
[truhlar.chem.umn.edu/content/book-chapters-pdf-files](http://truhlar.chem.umn.edu/content/book-chapters-pdf-files)

### ***Reading for Friday, January 26, 2018***

“An introduction to group theory for chemists”

J. E. White, *J. Chem. Educ.* **44**, 1280135 (1967).  
[doi.org/10.1021/ed044p128](https://doi.org/10.1021/ed044p128)

### ***Reading for Monday, January 29, 2018***

“Relationships between E2 and E1cB mechanisms of  $\beta$ -elimination,”

R. A. More O’Ferrall, *J. Chem. Soc. B*, 1970, 274-277.  
[doi.org/10.1039/J29700000274](https://doi.org/10.1039/J29700000274)

### ***Reading for Friday, February 2, 2018***

“Diatomic Molecules According to the Wave Mechanics. II. Vibrational Levels”

P M. Morse, *Phys. Rev.* 1929, 34, 57–64  
[doi.org/10.1103/PhysRev.34.57](https://doi.org/10.1103/PhysRev.34.57)

### ***Reading for Monday, February 5, 2018***

“The Relative Likelihood of Encountering Conical Intersections and Avoided Intersections on the Potential Energy Surfaces of Polyatomic Molecules,”

D. G. Truhlar and C. A. Mead, *Phys. Rev. A* **68**, 32501/1-032501/2 (2003).  
[doi.org/10.1103/PhysRevA.68.032501](https://doi.org/10.1103/PhysRevA.68.032501)

### ***Reading for Friday, February 9, 2018***

“Perspective on "Zur Quantentheorie der Molekeln",”

J. C. Tully, *Theor. Chem. Acc.* **103**, 173-176 (2000).  
[doi.org/10.1007/s002149900049](https://doi.org/10.1007/s002149900049)

### ***Readings for Monday, February 12, 2018 (three readings)***

“The Concept of Resonance,”

D. G. Truhlar, *Journal of Chemical Education* **84**, 781-782 (2007).  
[doi.org/10.1021/ed084p781](https://doi.org/10.1021/ed084p781)

“Are Molecular Orbitals Delocalized?”

D. G. Truhlar, *Journal of Chemical Education* **89**, 573-574 (2012).  
[doi.org/10.1021/ed200565h](https://doi.org/10.1021/ed200565h)

“Exploring the Nature of the H<sub>2</sub> Bond. 1. Using Spreadsheet Calculations To Examine the Valence Bond and Molecular Orbital Methods,” *Journal of Chemical Education* **90**, 1452-1358 (2012).

A. M. Halpern and E. D. Glendening  
[doi.org/10.1021/ed400234g](https://doi.org/10.1021/ed400234g)

**Reading for Friday, February 16, 2018**

“Nonadiabatic Transitions Between Ionic and Covalent States,”

R. K. Janev, *Adv. At. Mol. Phys.* **12**, 1-37 (1976). Read Sections I-III (pages 1-20).  
doi.org/10.1016/S0065-2199(08)60041-X

**Reading for Monday, February 19, 2018**

“Chemical Equilibrium in Relation to Chemical Kinetics,”

K. Denbigh, in *The Principles of Chemical Equilibrium* (Cambridge University Press, 1955),  
Chaper 15.

<https://ia800406.us.archive.org/30/items/ThePrinciplesOfChemicalEquilibrium/Denbigh-ThePrinciplesOfChemicalEquilibrium.pdf>

**Readings for Friday, February 23, 2018 (two readings)**

“The Crossing of Potential Energy Surfaces,”

E. Teller, *J. Phys. Chem.* **41**, 109-116 (1937).

<https://pubs.acs.org/doi/pdf/10.1021/j150379a010>

“Internal Conversion in Polyatomic Molecules,”

E. Teller, *Israel J. Chem.* **7**, 227-236 (1969).

<http://onlinelibrary.wiley.com/doi/10.1002/ijch.196900034/abstract> (Read whole pdf.)

**Reading for Monday, February 26, 2018**

“Molecular Dynamics with Electronic Transitions,”

J. C. Tully, *J. Chem. Phys.* **93**, 1061-1071 (1990).

doi.org/10.1063/1.459170 (Read sections I-III and V.)

**Readings for Friday, March 2, 2018 (two readings)**

“Activated Complex Theory of Bimolecular Reactions,”

B. H. Mahan, *Journal of Chemical Education* **51**, 709-711 (1974).

[dx.doi.org/10.1021/ed051p709](https://doi.org/10.1021/ed051p709)

“Perspective on ‘‘On the Theory of Oxidation-Reduction Reactions Involving Electron Transfer. I’’,”

J. M. Lluch, *Theor. Chem. Acc.* **103**, 231-233 (2000).

<https://link.springer.com/content/pdf/10.1007%2Fs002149900016.pdf>

**Readings for Monday, March 5, 2018 (three readings)**

“Vignette: Electron Transfer Reactions,”

in *Physical Chemistry*, 2<sup>nd</sup> ed., edited by R. S. Berry, S. A. Rice, and J. Ross,  
Oxford University Press, New York, 2000), pp. 945-948.

[https://comp.chem.umn.edu/courses/Marcus\\_Vignette.pdf](https://comp.chem.umn.edu/courses/Marcus_Vignette.pdf)

“Free Energy Surfaces for Liquid-Phase Reactions and Their Use to Study the Border Between  
Concerted and Nonconcerted  $\alpha,\beta$ -Elimination Reactions of Esters and Thioesters,”

Y. Kim, J. R. Mohrig, and D. G. Truhlar, *J. Am. Chem. Soc.* **131**, 11071-11082 (2010).

[dx.doi.org/10.1021/ja101104q](https://doi.org/10.1021/ja101104q) (Read pages 11071-11074.)

“Chemical Kinetics and Mechanisms of Complex Systems: A Perspective on Recent Theoretical  
Advances,”

S. J. Klippenstein, V. Pande, and D. G. Truhlar, *J. Am. Chem. Soc.* **136**, 528-546 (2014).

[dx.doi.org/10.1021/ja408723a](https://doi.org/10.1021/ja408723a)