

***Supporting Information for:***

**SM6T: A model to predict aqueous free energies of solvation as a function of temperature**

***Adam C. Chamberlin, Christopher J. Cramer,\* and Donald G. Truhlar\****

*Department of Chemistry and Supercomputing Institute, 207 Pleasant St. SE, University of Minnesota, Minneapolis, MN 55455-0431*

\*cramer@chem.umn.edu, [truhlar@umn.edu](mailto:truhlar@umn.edu)

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**Table S1: The compounds in the database, the number of experimental points used for each compound, the minimum and maximum temperatures for which experimental data was available, and the minimum and maximum experimental free energies of solvation for each compound**

	Number of Points	Minimum Experimental Temperature	Maximum Experimental Temperature	Minimum Experimental $\Delta G_S^\circ$	Maximum Experimental $\Delta G_S^\circ$
methane	10	285.05	348.35	1.82	2.75
ethane	23	273.15	345.55	1.29	2.67
propane	14	283.15	347.25	1.61	2.91
butane	17	276.15	349.25	1.45	3.20
pentane	13	283.15	372.25	1.85	3.90
hexane	22	273.15	372.25	1.43	4.20
heptane	15	273.15	372.25	1.48	4.74
octane	12	288.15	372.25	1.47	5.40
nonane	7	288.15	372.25	2.37	5.77
2-methylbutane	6	273.15	333.15	1.48	2.97
2,2-dimethylpropane	4	298.15	353.15	2.50	4.87
2-methylpentane	11	273.15	372.25	1.55	4.24
2,3-dimethylbutane	10	273.15	372.25	1.34	3.98
2,2-dimethylhexane	9	273	313	1.85	3.53
2,5-dimethylhexane	9	273	313	1.91	3.46
cyclopropane	8	293.15	318.15	0.69	1.18
cyclopentane	12	298	372.25	1.13	2.55
cyclohexene	10	277.95	318.35	-0.33	0.65
cyclohexane	7	287.95	318.35	0.66	1.95
methylcyclohexane	8	298.15	372.25	1.62	3.35
cyclooctene	9	273	313	-0.35	0.80
cyclooctane	9	273	313	0.12	1.50
ethylcyclohexane	9	273	313	0.83	2.23
cis-1,2-dimethylcyclohexane	9	273	313	0.96	2.36
trans-1,2-dimethylcyclohexane	10	273	313	0.63	2.05
trans-decalin	5	283.15	303.15	0.62	1.25
ethyne	4	273.15	333.15	-0.30	0.41
ethene	16	273	346.05	0.87	1.98
propyne	6	294.15	361.15	-0.39	0.96
propene	14	273.15	323.15	0.40	2.76
2-methyl-1,3-butadiene	4	293.15	333.15	0.66	1.23
benzene	78	275	373.14	-1.45	0.07
toluene	53	273	353	-1.60	-0.13
ethenylbenzene	10	280.15	338.15	-1.81	-0.62
o-xylene	28	273.15	353	-1.65	0.01
m-xylene	24	273	340.85	-1.73	-0.13
p-xylene	43	273.15	373.14	-1.63	0.57

	Number of Points	Minimum Experimental Temperature	Maximum Experimental Temperature	Minimum Experimental $\Delta G_S^\circ$	Maximum Experimental $\Delta G_S^\circ$
ethylbenzene	83	273.15	373.14	-1.58	0.60
propylbenzene	21	283	358.95	-1.03	0.77
cumene	12	298.09	353.39	-0.60	0.64
1,2,4-trimethylbenzene	7	288	318	-1.06	-0.29
1,3,5-trimethylbenzene	20	283	373.14	-0.98	0.73
naphthalene	9	276.7	308.5	-3.01	-2.13
tetralin	5	283.15	303.15	-1.93	-1.34
butylbenzene	14	280.15	373.14	-1.06	1.00
1-methylnaphthalene	12	277.1	313.15	-3.08	-2.06
2-methylnaphthalene	5	277.1	304	-2.85	-2.14
acenaphthalene	5	277.1	304	-3.78	-2.93
anthracene	4	284	304	-4.14	-3.45
phenanthrene	11	277.1	308.5	-4.29	-3.54
1-methylphenanthrene	5	277.1	304	-4.01	-3.58
pyrene	5	277.1	304	-4.72	-4.13
fluoranthene	5	283	328	-5.12	-3.86
water	94	274.15	373.15	-4.44	-2.53
methanol	11	273	357.75	-5.49	-4.30
ethanol	10	273	333.15	-5.64	-4.20
1,2-ethanediol	6	297.5	348.1	-9.35	-8.39
1-propanol	5	273	313.15	-5.59	-4.25
2-propanol	11	273	373.15	-5.55	-3.03
1-butanol	58	273	373.15	-5.51	-3.02
2-butanol	26	273	363.35	-5.59	-3.14
2-methyl-1-propanol	26	273.15	372.19	-5.32	-2.92
cyclopentanol	8	273.15	363.65	-6.40	-3.92
1-pentanol	19	273.15	373.15	-5.53	-2.58
2-pentanol	9	298.15	363.45	-4.32	-2.47
3-pentanol	13	273.15	363.15	-5.16	-2.43
2-methyl-1-butanol	13	273.65	363.95	-5.46	-2.58
2-methyl-2-butanol	13	273.65	363.35	-4.99	-2.39
2,2-dimethyl-1-propanol	12	285.15	363.15	-4.37	-1.93
3-methyl-1-butanol	18	283.25	363.15	-4.90	-2.63
3-methyl-2-butanol	11	273.15	363.15	-5.21	-2.36
phenol	15	277.15	373.15	-6.79	-4.95
cyclohexanol	11	303.75	363.45	-5.33	-3.71
1-hexanol	32	273.15	373.15	-5.40	-2.27
2-hexanol	12	273.15	363.35	-5.19	-2.10
3-hexanol	13	273.15	363.45	-5.19	-2.09
4-methyl-2-pentanol	11	273.15	363.35	-5.06	-2.03
2-ethyl-1-butanol	10	273.15	363.65	-5.39	-2.21
benzylalcohol	12	273.15	328.15	-7.61	-5.89
3-methylphenol	26	283.15	371.65	-6.26	-4.71
2-methylphenol	21	278.25	373.15	-6.31	-4.27
4-methylphenol	16	278.2	373.15	-6.77	-4.69

	Number of Points	Minimum Experimental Temperature	Maximum Experimental Temperature	Minimum Experimental $\Delta G_S^\circ$	Maximum Experimental $\Delta G_S^\circ$
cycloheptanol	8	293.15	363.75	-5.77	-3.59
2-methylcyclohexanol	9	273.15	353.25	-5.47	-3.04
3-methylcyclohexanol	9	273.15	363.85	-5.85	-3.02
4-methylcyclohexanol	10	273.15	363.65	-6.10	-3.09
cyclohexylmethanol	7	303.05	363.35	-4.09	-2.47
1-heptanol	18	279.19	373.15	-5.03	-1.96
2-heptanol	9	283.35	363.35	-4.63	-1.71
3-heptanol	9	273.15	363.35	-5.17	-1.63
4-heptanol	9	282.65	363.85	-4.75	-1.69
2,4-dimethyl-3-pentanol	6	312.65	363.75	-2.86	-1.33
2-phenylethanol	6	283.15	333.45	-6.75	-5.51
cyclooctanol	8	283.15	363.65	-6.21	-3.60
1-octanol	5	293.96	342.46	-4.17	-1.86
2-octanol	12	273.15	363.65	-5.25	-1.52
3-octanol	9	283.15	363.25	-4.56	-1.41
2-ethyl-1-hexanol	7	302.75	363.45	-3.64	-1.68
1-nonanol	9	282.95	363.65	-5.26	-1.56
2-nonanol	10	273.15	363.65	-5.23	-1.14
formaldehyde	9	283	318.15	-6.94	-6.29
acetaldehyde	10	273	313.15	-4.03	-3.06
propanal	5	283	318.15	-3.82	-2.93
2-oxo-propanal	4	288	318.15	-6.98	-6.27
butanal	6	283	318.15	-3.75	-1.97
pentanal	6	283	318	-3.63	-1.15
furfural	42	273.15	371.05	-5.79	-4.42
benzaldehyde	18	273.15	363.15	-4.84	-3.10
hexanal	10	273.15	363.15	-3.60	-0.98
heptanal	12	273.15	363.15	-3.66	-0.82
octanal	5	283	318	-3.33	-1.56
nonanal	4	283	318	-3.00	-0.95
2-methylprop-2-enal	11	278	333.15	-3.24	-2.09
acetone	13	273.66	333.15	-4.38	-3.17
2-butanone	23	283	373.15	-4.55	-2.46
cyclopentanone	10	273.15	363.85	-5.83	-4.31
2-pentanone	25	273.15	363.65	-4.11	-2.04
3-pentanone	16	273.15	353.35	-4.07	-2.15
3-methyl-2-butanone	10	273.15	362.15	-4.06	-1.92
isopropylacetone	9	273.15	348.15	-3.53	0.15
cyclohexanone	10	273.15	363.85	-5.74	-3.70
2-hexanone	15	273.15	364.65	-4.13	-1.76
4-methyl-2-pentanone	18	273.15	363.55	-3.87	-1.49
3,3-dimethylbutanone	12	282.65	363.35	-3.18	-1.22
2-heptanone	14	273.15	363.65	-3.99	-1.34
4-heptanone	15	273.15	363.65	-3.99	-1.34
5-methyl-2-hexanone	10	273.15	362.85	-3.90	-1.33

	Number of Points	Minimum Experimental Temperature	Maximum Experimental Temperature	Minimum Experimental $\Delta G_S^\circ$	Maximum Experimental $\Delta G_S^\circ$
2,4-dimethyl-3-pentanone	10	273.15	363.45	-2.64	-0.06
2-octanone	9	273.15	364.15	-4.67	-1.06
acetophenone	7	288	318	-5.21	-4.09
2,6-dimethyl-4-heptanone	10	273.15	363.85	-3.15	0.03
1-methoxypropane	6	273.15	298.15	-2.35	-1.64
tetrahydrofuran	5	293.15	343.15	-3.58	-2.72
2-methoxypropane	5	283.15	298.15	-2.32	-1.90
ethoxyethane	34	273.15	355.15	-2.72	-0.66
2-methyltetrahydrofuran	8	273.15	343.75	-4.09	-2.26
2-methoxy-2-methylpropane	9	273.15	323.15	-2.92	-1.20
1-propoxypropane	4	273.15	298.15	-2.10	-1.17
2,5-dimethyltetrahydrofuran	9	273.15	356.15	-3.84	-1.25
1-ethoxybutane	10	273.15	363.85	-2.70	-0.13
2-Isopropoxypropane	7	273.15	334.15	-2.42	0.20
methoxybenzene	8	283.35	363.85	-3.19	-1.44
1-butoxybutane	10	273.15	363.65	-2.11	1.58
methylethanoate	14	278.15	345.05	-3.68	-2.95
ethylformate	10	278.15	348.65	-2.93	-2.25
methylpropanoate	14	273.15	343.35	-3.60	-2.10
ethylethanoate	32	273.15	343.65	-3.64	-2.16
propylformate	7	277.15	318.15	-3.02	-2.13
isobutylformate	7	273.15	333.55	-2.84	-1.36
propylethanoate	14	273.15	363.35	-3.59	-1.23
2-propylethanoate	9	273.15	347.75	-3.32	-1.43
1-methylbutanoate	10	273.15	363.65	-3.49	-1.21
1-ethylpropanoate	9	273.15	353.45	-3.54	-1.49
diethylcarbonate	10	273.15	363.45	-4.10	-1.85
dimethylmaleate	10	273.15	363.65	-7.16	-5.30
diethyloxalate	6	273.15	322.85	-6.43	-4.73
1-ethylbutanoate	10	273.15	363.65	-3.28	-0.76
ethyl-2-methylproanoate	10	273.15	363.75	-3.01	-0.41
1-butylethanoate	13	273.15	363.65	-3.50	-0.96
2-methylpropylethanoate	8	273.15	353.35	-3.26	-1.06
2-butylethanoate	9	273.15	363.65	-3.25	-0.80
tert-butylethanoate	9	273.15	353.65	-2.72	-0.49
pentylformate	10	273.15	363.65	-3.39	-0.75
1-propylpropanoate	10	273.15	363.65	-3.39	-0.67
methylbenzoate	9	273.15	363.65	-4.70	-2.56
ethyl 3-methylbutanoate	10	273.15	363.65	-3.05	-0.29
1-propylbutanoate	8	273.15	363.65	-3.19	-0.53
pentylethanoate	8	273.15	353.25	-3.53	-1.00
3-methylbutylethanoate	10	273.15	363.85	-3.30	-0.67
methylsalicylate	9	273.15	363.65	-6.29	-2.79
diethylsuccinate	10	273.15	363.85	-7.43	-4.46
1-butylbutyrate	10	273.15	363.45	-3.40	-0.17

	Number of Points	Minimum Experimental Temperature	Maximum Experimental Temperature	Minimum Experimental $\Delta G_S^\circ$	Maximum Experimental $\Delta G_S^\circ$
2-methylpropyl-2- methylpropanoate	10	273.15	363.35	-4.63	-2.67
ethylbenzoate	9	273.15	363.45	-4.42	-2.05
octylmethanoate	8	282.35	363.65	-3.43	-0.87
formic acid	4	277.78	308.64	-7.09	-6.74
ethanoic acid	4	277.78	308.64	-7.20	-6.68
pentanoic acid	9	277.78	308.64	-6.92	-6.08
hexanoic acid	13	273.15	333.15	-7.04	-5.16
heptanoic acid	9	273.15	333.15	-7.73	-5.17
octanoic acid	15	273.15	373.15	-6.98	-3.87
nonanoic acid	4	303.15	333.15	-5.64	-4.42

**Table S2: List of compounds for which experimental free energies of solvation were computed from Henry's Law constants, and the references and compilations of references from which they were obtained.**

	Sources for Henry's Law		
	Constants	Compilations	
2,2-dimethylhexane	1		
2,5-dimethylhexane	1		
cyclopentane	2	3	
cyclooctene	1		
cyclooctane	1		
ethylcyclohexane	1		
cis-1,2-dimethylcyclohexane	1		
trans-1,2-dimethylcyclohexane	1		
trans-decalin	4	3	
ethene	5		
benzene	5-13	3	14
toluene	4-9,12,13,15-20	3	14
o-xylene	4,7,9,19,20	3	
m-xylene	4,5,7,9,20	3	14
p-xylene	2,4,5,7,9,20	3	14
ethylbenzene	4,7-9,13,16,20	3	
propylbenzene	4,16,20	3	
1,2,4-trimethylbenzene	2,20	3	
1,3,5-trimethylbenzene	4,13,20	3	
naphthalene	10	3	
tetralin	4	3	
1-methylnaphthalene	21	3	
2-methylnaphthalene	21	3	
acenaphthalene	21	3	
anthracene	10,21	3	
phenanthrene	10,21	3	
1-methylphenanthrene	21	3	
pyrene	21	3	
fluoranthene	22	3	
water	23-27		
methanol	28		14
ethanol	28		14
1,2-ethanediol	29		
1-propanol	28		14
2-propanol	28		14
1-butanol	28		14
2-butanol	28		14
phenol	30,31		

	<u>Sources for Henry's Law Constants</u>	<u>Compilations</u>
3-methylphenol	31,32	
2-methylphenol	31,32	
4-methylphenol	31,32	
formaldehyde	33,34	14
acetaldehyde	28,33,35	14
propanal	34	14
2-oxo-propanal	33	14
butanal	34	14
pentanal	34	14
benzaldehyde	34,36	14
heptanal	34	14
octanal	34	14
nonanal	34	14
2-methylprop-2-enal	36	14
acetone	34,35,37	14
2-butanone	34	14
acetophenone	36,37	14
tetrahydrofuran	29	
ethoxyethane	15	14
2-methoxy-2-methylpropane	7	14
formic acid	38	14
ethanoic acid	38	14
pentanoic acid	38,39	14



**Table S3: List of compounds for which experimental free energies of solvation were computed from vapor pressures and solubilities, and the references and compilations of solubilities from which they were obtained.**

	Sources for Vapor Pressures	Sources for Solubilities	Compilations	
methane	40	41		
ethane	40	41-43		
propane	40	41,43	44	
butane	40	41,45	44	
pentane	40	46-48	44	
hexane	49	46-48,50,51	44	52
heptane	49	46-48,50	44	
octane	49	47,48,51	44	52
nonane	49	47,48,51	44	52
2-methylbutane	40	50,53	44	52
2,2-dimethylpropane	40	54	44	
2-methylpentane	49	47,48,50	44	
2,3-dimethylbutane	49	47,48,50	44	
cyclopropane	40	55	44	
cyclopentane	40	47,48	44	
cyclohexene	56	57	44	
cyclohexane	40	57	44	
methylcyclohexane	49	47,48	44	
ethyne	40	43	44	
ethene	40	41		
propyne	40	58	44	
propene	40	43,59	44	
2-methyl-1,3-butadiene	40	53		52
benzene	40	8,43,60-69	44	
ethenylbenzene	49	70	44	
o-xylene	49	50,71	44	
m-xylene	49	50,72	44	
p-xylene	49	50,71-73	44	
ethylbenzene	49	8,13,50,71,73-76	44	
propylbenzene	49	13,75	44	
cumene	49	77	44	
1,3,5-trimethylbenzene	49	73	44	
butylbenzene	49	73,76	44	
1-methylnaphthalene	49	78,79	44	
1-butanol	40,80	81-88	44	

	Sources for Vapor Pressures	Sources for Solubilities	Compilations
2-butanol	40	81,89-91	44
2-methyl-1-propanol	40,80	92	44
cyclopentanol	40	81	
1-pentanol	40	84,93,94	44
2-pentanol	40	93,94	44
3-pentanol	40	93,94	44
2-methyl-1-butanol	49	93,94	44
2-methyl-2-butanol	40	93,94	44
2,2-dimethyl-1-propanol	40	93,94	44
3-methyl-1-butanol	40	93-96	44
3-methyl-2-butanol	80	93,94	44
cyclohexanol	40,80	81,97,98	44
1-hexanol	40	84,85,94,99	44
2-hexanol	40	94,100	44
3-hexanol	40	94,100	44
4-methyl-2-pentanol	40	94,100	44
2-ethyl-1-butanol	49	81	
benzylalcohol	49	81,101	44
3-methylphenol	49,80	102-105	44
2-methylphenol	49	97,105	44
cycloheptanol	40	81	
2-methylcyclohexanol	49	81	
3-methylcyclohexanol	49	81	
4-methylcyclohexanol	49	81	
cyclohexylmethanol	49	81	
1-heptanol	40	84,85,94,99	44
2-heptanol	40	94	44
3-heptanol	40	81	
4-heptanol	40	81	
2,4-dimethyl-3-pentanol	80	94	44
2-phenylethanol	49	81	
cyclooctanol	40	81	
1-octanol	40	106	44
2-octanol	40,49	43,81,107	44
3-octanol	40	81	
2-ethyl-1-hexanol	40	94	44
1-nonanol	49	81	
2-nonanol	49	81	
butanal	56	108	
furfural	56	108-113	44
benzaldehyde	56	108	
hexanal	56	108	
heptanal	56	108	
2-methylprop-2-enal	56	108	

	Sources for Vapor Pressures	Sources for Solubilities	Compilations
2-butanone	40	43,114,115	44
cyclopentanone	56	116	
2-pentanone	56	114-118	44
3-pentanone	56	115,116,118	44
3-methyl-2-butanone	56	116	
isopropylacetone	49	115,118,119	44
cyclohexanone	56	116	
2-hexanone	56	115,116,118	44
4-methyl-2-pentanone	56	115,116,118,119	44
3,3-dimethylbutanone	56	115,116	44
2-heptanone	56	115,116	44
4-heptanone	56	116,118	44
5-methyl-2-hexanone	56	116	
2,4-dimethyl-3-pentanone	56	116	
2-octanone	56	116	
acetophenone			
2,6-dimethyl-4-heptanone	56	116	
1-methoxypropane	40	120	44
tetrahydrofuran	49		
2-methoxypropane	40	120	44
ethoxyethane	40	43,96,121-124	44
2-methyltetrahydrofuran	56	116	
2-methoxy-2-methylpropane	56	116	
1-propoxypropane	40	120	44
2,5-dimethyltetrahydrofuran	56	116	
1-ethoxybutane	56	116	
2-Isopropoxypropane	56	116	
methoxybenzene	56	116	
1-butoxybutane	56	116	
methylethanoate	40	125,126	44
ethylformate	40	125	44
methylpropanoate	40,49	81,125	44
ethylethanoate	49	81,125,127-131	44
propylformate	49	125	44
isobutylformate	49	81	
propylethanoate	40,49	81	
2-propylethanoate	40,49	81	
1-methylbutanoate	49	81	
1-ethylpropanoate	49	81	
diethylcarbonate	49	81	
dimethylmaleate	49	81	
diethyloxalate	49	81	

	Sources for Vapor Pressures	Sources for Solubilities	Compilations
1-ethylbutanoate	49	81	
ethyl-2-methylproanoate	49	81	
1-butylethanoate	40,49	81	
2-methylpropylethanoate	49	81	
2-butylethanoate	49	81	
tert-butylethanoate	49	81	
pentylformate	49	81	
1-propylpropanoate	49	81	
methylbenzoate	49	81	
ethyl 3-methylbutanoate	49	81	
1-propylbutanoate	49	81	
pentylethanoate	49	81	
3-methylbutylethanoate	49	81	
methylsalicylate	49	81	
diethylsuccinate	49	81	
1-butylbutyrate	49	81	
2-methylpropyl-2- methylpropanoate	49	81	
ethylbenzoate	49	81	
octylmethanoate	49	81	
hexanoic acid	49	132-135	44
heptanoic acid	49	132,135	44
octanoic acid	49	53,132,135,136	44
nonanoic acid	49	136	44

**Table S4: List of compounds for which experimental free energies of solvation were computed from vapor pressures and activity coefficients at infinite dilution, and the references and compilations of solubilities from which they were obtained.**

	Sources for Vapor	Sources for Activity	Compilation
	Pressures	Coefficients at Infinite Dilution	
benzene	40	137	138
methanol	80	139,140	138
ethanol	80	139,141	138
1-propanol	80	139,141	138
2-propanol	80	140,142	138
1-butanol	40,80	143-145	138
2-methyl-1-propanol	40,80	15,143	
cyclohexanol	40,80	143	
3-methylphenol	49,80	146	138
acetone	40	144,147	138
2-butanone	40	143-145	138
tetrahydrofuran	49	143	
propylethanoate	40,49	146	138
2-propylethanoate	40,49	146	138
1-butylethanoate	40,49	146	138

**Table S5: Compounds for which experimental free energies of solvation at 298 K from the database used to parameterize SM6<sup>148</sup> were used.**

ethane	2-hexanone
pentane	2-heptanone
hexane	acetophenone
heptane	
octane	1-methoxypropane
	2-methoxypropane
2-methylbutane	ethoxyethane
2,2-dimethylpropane	
2-methylpentane	propylethanoate
	1-butylethanoate
cyclopropane	
cyclopentane	pentanoic acid
methylcyclohexane	hexanoic acid
cis-1,2-dimethylcyclohexane	heptanoic acid
	octanoic acid
	nonanoic acid
ethyne	
ethene	
propyne	
o-xylene	
m-xylene	
ethylbenzene	
naphthalene	
anthracene	
methanol	
ethanol	
1-propanol	
2-propanol	
1-butanol	
3-methylphenol	
acetaldehyde	
propanal	
butanal	
pentanal	
benzaldehyde	
octanal	
acetone	
2-butanone	
2-pentanone	
3-pentanone	

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