

## Supporting information for

**Tests of the RPBE, revPBE,  $\tau$ -HCTHhyb,  $\omega$ B97X-D, and MOHLYP  
density functional approximations and 29 others  
against representative databases  
for diverse bond energies and barrier heights in catalysis**

Ke Yang, Jingjing Zheng

*Department of Chemistry and Supercomputing Institute,  
University of Minnesota, 207 Pleasant Street S.E.,  
Minneapolis, MN 55455-0431, USA*

Yan Zhao

*Commercial Print Engine Labs, Hewlett-Packard Co.  
1501 Page Mill Road  
Palo, Alto, CA 94304*

and Donald G. Truhlar

*Department of Chemistry and Supercomputing Institute,  
University of Minnesota, 207 Pleasant Street S.E.,  
Minneapolis, MN 55455-0431, USA*

Date of finalization of supporting information: March, 09, 2010

Number of pages including this page: 16

This supporting information contains the comparison of SCF and post-SCF results for the SOGGA functional, the spin quantum numbers of ground states of V and Zr as calculated by each functional, extended versions of Table 2 to 5, and the gradient enhancement factors  $F_X$  as a function the reduced gradient densities  $s$  for the TPSS, M06-L, VS98, and  $\tau$ -HCTH meta-GGA functionals, and the expressions of exchange enhancement factors  $F_X$  of the GGA and meta-GGA exchange functionals in Figure 1 and Figure S1-S3.

The extended versions of Table 2 to 5 differ from those tables in the main text in two respects: (i) results are rounded to the hundredth of a kcal/mol rather than to the tenth of a kcal/mol; (ii) an additional functional, MOHLYP2, is included. Note that MOHLYP is the original MOHLYP functional of Ref. 4, and MOHLYP2 is a different version presented in Ref. 5; MOHLYP2 was mislabeled as MOHLYP in Ref. 5. MOHLYP2 gives a spin quantum number of 5/2 for the ground state of V and 2 for the ground state of Zr.

Note:

Some of the results differ quantitatively (not enough to change previous conclusions) from previously published numbers. The main sources of these differences are as follows: (i) In some previous work we used fine grids; in the present paper we always use ultrafine grids because we found that the results were not completely converged with fine grids. (ii) In some previous calculations (e.g., the BLYP calculations of the DBH24/08 paper<sup>5</sup>), we used the option SCF=(TIGHT,XQC), whereas in this article we used SCF=(XQC, MAXCONVENTIONALCYCLE=20,TIGHT,INTREP), which ensure the energies to converge to a threshold of  $10^{-6}$  Hartree. (iii) Some of the previous calculations were done with a version of the computer code that had a small ( $10^{-3}$ – $10^{-4}$  hartrees) bug in the SOGGA functional.

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Table S1. Comparison of the calculated bond energies (kcal/mol) by SCF calculations and post-SCF calculations with the SOGGA functional

	Bond energy <sup>a</sup>	Bond energy <sup>b</sup>
SCF (V <sub>2</sub> )	82.59	119.05
Post-SCF (V <sub>2</sub> ) <sup>c</sup>	82.59	119.05
SCF (VS)	121.53	139.75
Post-SCF (VS) <sup>c</sup>	121.55	139.79
SCF for (NiCH <sub>2</sub> <sup>+</sup> )	98.08	98.08
Post-SCF (NiCH <sub>2</sub> <sup>+</sup> ) <sup>c</sup>	98.07	98.07

<sup>a</sup>Bond energies calculated by using calculated atomic ground states (V,  $S = 5/2$ ; Ni<sup>+</sup>,  $S = 1/2$ )

<sup>b</sup>Bond energies calculated by using experimental atomic ground states (V,  $S = 3/2$ ; Ni<sup>+</sup>,  $S = 1/2$ ).

<sup>c</sup>Post-SCF calculations were done by using the PBEsol densities.

Table S2. Spin quantum numbers of ground states of V and Zr as calculated by various functionals

Functional	V	Zr
BP86	5/2	1
BVP86	5/2	1
BLYP	5/2	1
PW91	5/2	2
B3PW91	5/2	2
B3LYP	5/2	1
B3V5LYP	5/2	1
PBE	5/2	2
B98	3/2	1
revPBE	5/2	2
VS98	5/2	1
PBEhole	5/2	2
PBE0	5/2	2
RPBE	5/2	1
$\tau$ -HCTH	3/2	1
$\tau$ -HCTHhyb	3/2	1
TPSS	5/2	1
TPSSKCIS	5/2	1
TPSSH	5/2	1
BMK	5/2	1
TPSS1KCIS	5/2	1
MOHLYP	5/2	1
MOHLYP2	5/2	2
MPWLYP1M	5/2	1
B97-3	3/2	1
M05	3/2	1
WC06	5/2	2
B97-D	3/2	1
M06-L	3/2	1
LC- $\omega$ PBE	5/2	2
M06	3/2	1
PBESol	5/2	2
SOGGA	5/2	2
$\omega$ B97X-D	5/2	1
HSE	5/2	2

Table S3. Mean signed and unsigned errors in bond energies (kcal/mol per bond)<sup>a,b</sup>

	MPWLYPIM	0.09	0.97	-8.56	8.56	1.39/3.37	1.87/3.85	8.06/9.05	8.06/9.05	0.56	4.76
B97-3	-0.18	0.57	-4.63	4.63	-17.25	17.25	-3.11	8.07	8.07	-5.61	6.85
M05	-0.02	0.40	-5.78	5.78	-5.91	8.29	-1.62	4.93	4.93	-2.96	4.36
WC06 <sup>e</sup>	5.17	5.54	1.48	2.34	8.39/19.84	8.39/19.84	17.20/22.00	17.20/22.00	17.20/22.00	9.54	9.86
B97-D	-0.11	0.38	-7.66	7.66	17.93	17.93	7.42	7.42	7.42	3.89	7.46
M06-L	0.49	0.72	-5.54	5.54	0.52	4.22	7.01	7.01	7.01	0.61	3.97
LC- $\omega$ PBE	-0.41	0.87	-4.70	4.70	-33.55/-26.46	33.55/26.46	-8.07/-4.80	10.09/6.82	10.09/6.82	-9.28	9.88
M06	0.43	0.57	-1.89	1.89	-8.92	8.92	-1.35	4.35	4.35	-2.56	3.56
PBEsol	7.25	7.25	4.15	4.15	13.6/22.93	13.6/22.93	19.41/23.62	19.41/23.62	19.41/23.62	12.18	12.18
SOGGA	6.97	7.26	5.22	5.22	10.76/21.43	11.79/22.46	19.91/24.46	19.91/24.46	19.91/24.46	11.99	12.31
$\omega$ B97X-D	-0.17	0.41	-2.15	2.15	-20.48/-19.84	20.84/19.84	-2.54/-2.22	6.14/5.83	6.14/5.83	-5.54	6.42
HSE	-0.56	0.88	-5.66	5.66	-21.95/-15.12	21.95/15.12	-4.72/-1.53	7.96/5.56	7.96/5.56	-6.26	7.17

a Basis set: Metals: TZQ; nonmetals: MG3 in MII BE4 MG3S in AE6 and 6-311+G(3df 2p) in ABDE4

*b* A positive MSE corresponds to the functional predicting too large of a bond strength. Note that in computing the mean errors per bond for AE6, as explained in ref. 2, we first compute the mean error in the atomization energy, and then we divide by 4.83, which is the mean number of bonds per molecule. It is not necessary to divide by the average number of bonds for TMAE4 because those data are for diatomics which have only a single bond. MLBE4 has three cases where a single bond is broken and one case where five bonds are broken; for that data set, as explained in ref. 4, we divide the error for  $\text{Fe}(\text{CO})_5$  by 5 before we compute the mean errors for MLBE4.

calculated with the calculated atomic experimental atomic ground states

The bond dissociation energies of molecules in TMAE4 and MLBE4 databases were calculated in two ways if the calculated atomic ground states are different than the experimental atomic ground states. The numbers before the slash “/” are the MSEs and MUEs calculated with the calculated atomic ground states; while the numbers after the slash are the MSEs and MUEs calculated with the experimental atomic ground states.

<sup>d</sup> AMSE and AMUE are the average of MSEs and MUEs for AE6, ABDE4, TMAE4, and MLBE4 databases weighted 6/18:4/18:4/18; for TMAE4 and MLBE4 databases, the MSEs and MUEs used in AMSE and AMUE are chosen as the average of the pre-slash and post-slash values. As explained in footnote *b*, the mean errors averaged to obtain the final of this table are all on a per bond basis before we average them.

<sup>e</sup> The bond dissociation energy of V<sub>2</sub>, NiCH<sub>2</sub><sup>+</sup>, and VS were obtained by post-SCF calculations with PBEsol densities of convergence problems for V and Ni<sup>+</sup> atoms.

Table S4. Mean signed errors and mean unsigned errors in barrier heights (kcal/mol)<sup>a</sup>

B97-3	-2.29	2.49	-0.48	0.96	0.71	1.63	-2.16	2.20	-1.06	1.82
M05	-3.26	4.80	0.01	0.95	1.00	2.48	-0.63	1.74	-0.72	2.49
WC06	-15.70	15.70	-7.58	7.58	-3.36	3.36	-11.53	11.53	-9.54	9.54
B97-D	-9.40	9.40	-6.19	6.19	-1.25	1.71	-6.01	6.39	-5.71	5.92
M06-L	-5.85	6.87	-3.35	3.35	0.52	1.77	-4.11	4.21	-3.20	4.05
LC- $\omega$ PBE	1.07	2.42	2.82	2.82	1.30	2.09	-1.01	1.15	1.05	2.12
M06	-3.62	4.06	-1.61	1.64	0.54	1.91	-1.53	1.66	-1.56	2.32
PBEsol	-17.50	17.50	-7.40	7.40	-3.90	3.90	-12.69	12.69	-10.37	10.37
SOGGA	-17.04	17.04	-7.14	7.14	-3.87	3.87	-12.85	12.85	-10.23	10.23
$\omega$ B97X-D	-1.31	2.19	0.10	0.67	1.03	2.04	-2.04	2.24	-0.56	1.79
HSE	-6.10	6.10	-2.58	2.58	-0.65	1.87	-4.56	4.56	-3.47	3.78

*a* Single-point calculations with MG3S basis set with QCISD/MG3 geometries.

*b* AMSE and AMUE are the average of MSEs and MUEs for the HATBH6, NSBH6, UABH6, and HTBH6 databases.

Table S5. Average error for catalytical energies (AECE) of various functionals, weighting DBE18 and DBH24 equally

Functional	AECE
M06	2.94
M05	3.42
M06-L	4.01
$\tau$ -HCTHhyb	4.09
$\omega$ B97X-D	4.10
B97-3	4.33
B98	4.72
VS98	4.77
TPSS1KCIS	5.14
B3PW91	5.18
PBE0	5.23
B3LYP	5.43
HSE	5.47
B3V5LYP	5.60
BMK	5.75
LC- $\omega$ PBE	6.00
MOHLYP	6.05
MPWLYP1M	6.08
TPSSh	6.16
RPBE	6.50
TPSSKCIS	6.58
revPBE	6.60
B97-D	6.69
TPSS	7.16
$\tau$ -HCTH	7.25
BLYP	7.33
BVP86	7.84
PBE	7.84
PW91	7.98
PBEhole	8.04
BP86	8.19
WC06	9.70
SOGGA	11.27
PBESol	11.28
MOHLYP2	12.96

Table S6. Mean signed errors and mean unsigned errors in barrier heights for two basis sets (kcal/mol)

Method	HATBH6			NSBH6			UABH6			HTBH6			DBH24/08	
	MSE	MUE	MSE	MUE	MSE	MUE	MSE	MUE	MSE	MUE	MSE	MUE	AMUE	
MG3S														
M06-2X	-0.02	0.73	0.60	0.86	0.37	1.09	-0.49	1.24	0.98					
M08-SO	-0.57	1.08	0.19	0.69	0.23	1.43	-0.81	1.00	1.05					
BMK	-0.96	0.96	0.60	0.77	0.97	2.05	-1.12	1.12	1.22					
$\omega$ B97X-D	-1.31	2.19	0.10	0.67	1.03	2.04	-2.04	2.24	1.78					
B97-3	-2.29	2.49	-0.48	0.96	0.71	1.63	-2.16	2.20	1.82					
LC- $\omega$ PBE	1.07	2.42	2.82	2.82	1.30	2.09	-1.01	1.15	2.12					
M06	-3.62	4.06	-1.61	1.64	0.54	1.91	-1.53	1.66	2.32					
maug-cc-pV(T+d)Z														
M06-2X	-0.20	0.41	0.84	0.84	0.19	1.13	-0.50	1.26	0.91					
M08-SO	-0.62	0.68	0.80	0.80	0.15	1.32	-0.77	0.91	0.93					
BMK	-1.44	1.44	0.72	0.94	0.80	1.93	-1.07	1.47	1.45					
$\omega$ B97X-D	-1.34	1.73	0.73	0.82	0.89	1.98	-1.87	2.31	1.71					
B97-3	-2.65	2.69	-0.14	1.30	0.48	1.56	-2.16	2.27	1.95					
LC- $\omega$ PBE	1.01	2.22	3.43	3.43	1.12	2.10	-0.90	1.16	2.23					
M06	-3.55	3.99	-0.91	1.30	0.47	1.76	-1.36	1.69	2.18					

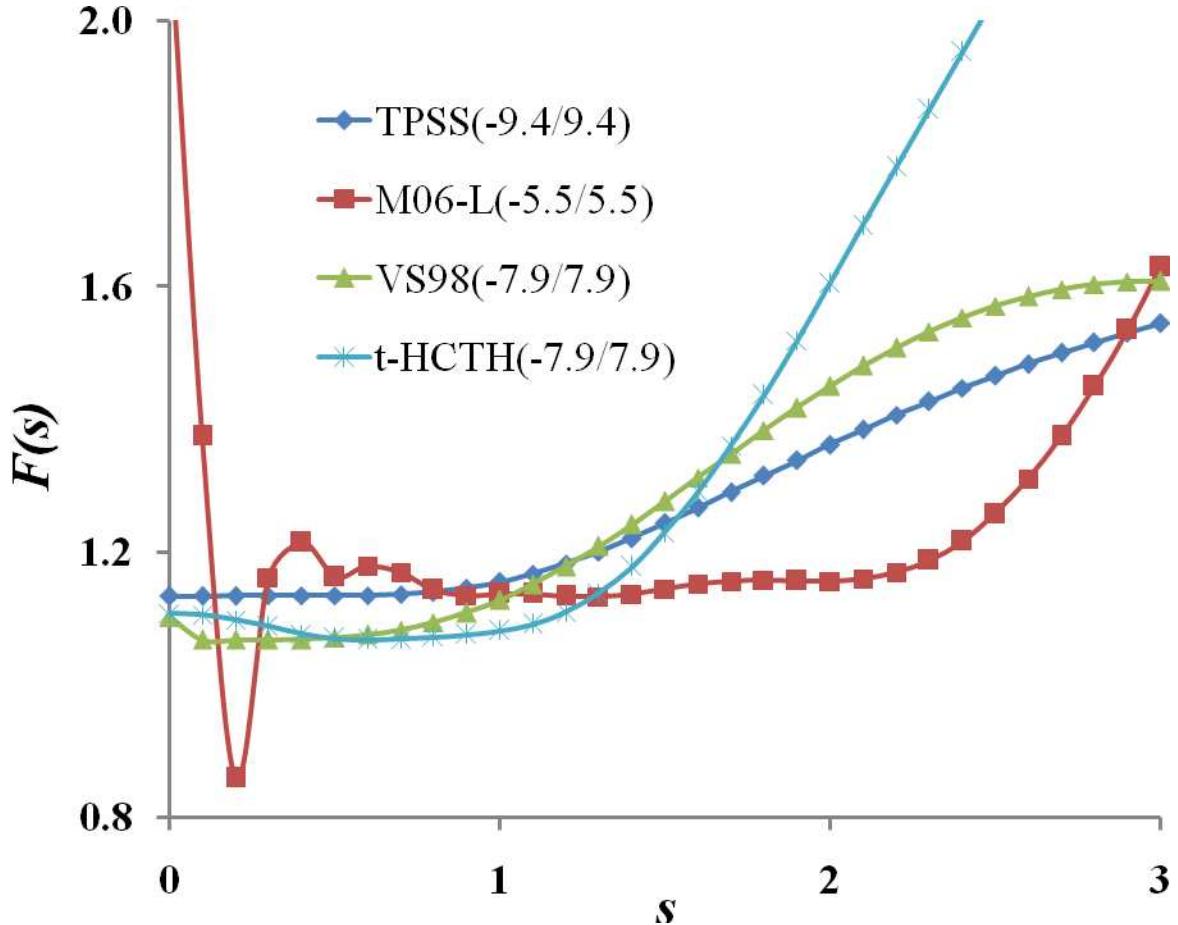


Figure S1. Exchange enhancements factors  $F_X$  as function of reduced gradient  $s$  ( $\alpha = 0$ ). (the numbers in the parentheses are the MSEs and MUEs of meta-GGA functionals for the ABDE4 database in units of kcal/mol)

Note:  $s$  and  $\alpha$  are two dimensionless variables defined as following:

$$s = \frac{|\nabla n|}{2(3\pi^2)^{1/3} n^{4/3}}$$

$$\alpha = \frac{\tau - \tau^W}{\tau^{LDA}}$$

where  $n$  is the electron density,  $\tau$  is the kinetic energy density,  $\tau^W$  is the von Weizsäcker kinetic energy density, and  $\tau^{LDA}$  is the kinetic energy density of uniform electron gas.

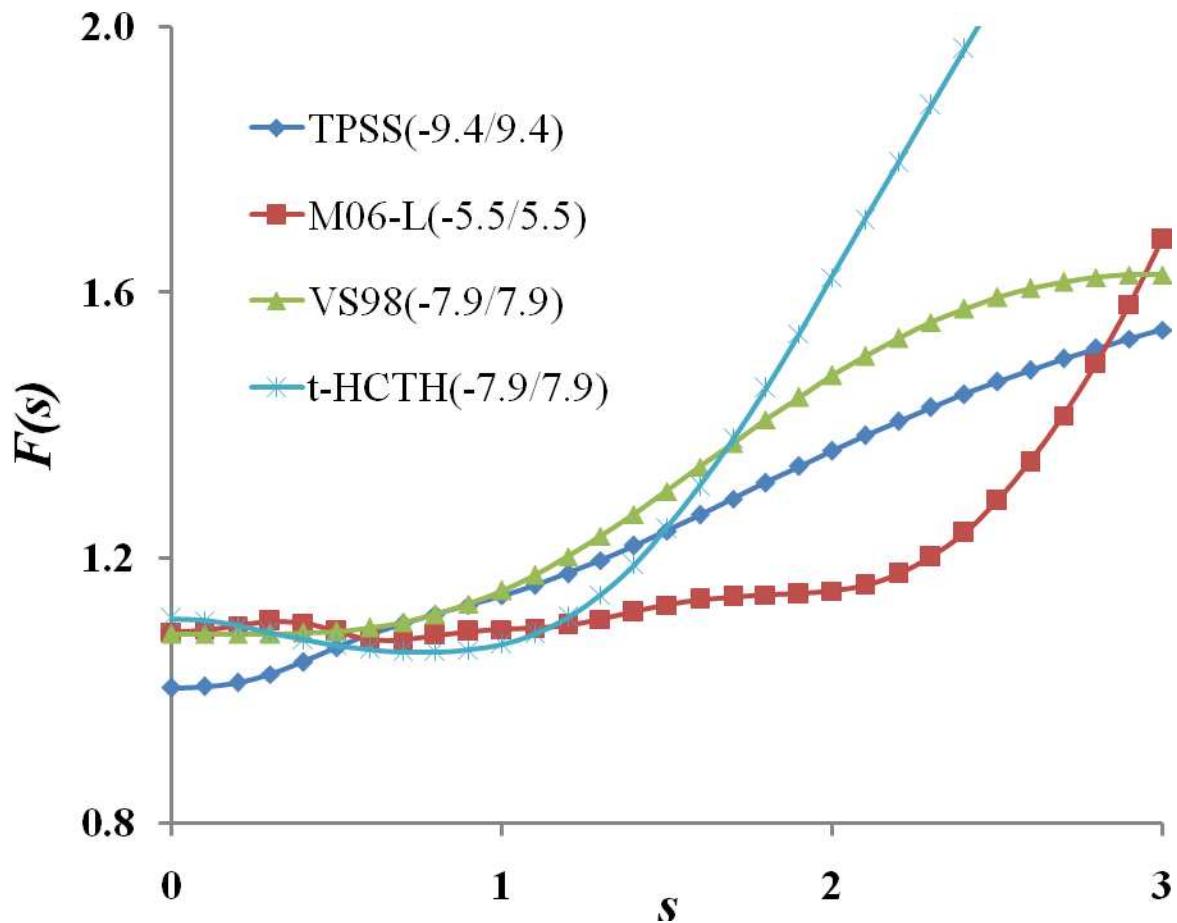


Figure S2. Exchange enhancement factors  $F_X$  as function of reduced gradient  $s$  ( $\alpha = 0.5$ ). (the numbers in the parentheses are the MSEs and MUEs of meta-GGA functionals for the ABDE4 database in units of kcal/mol)

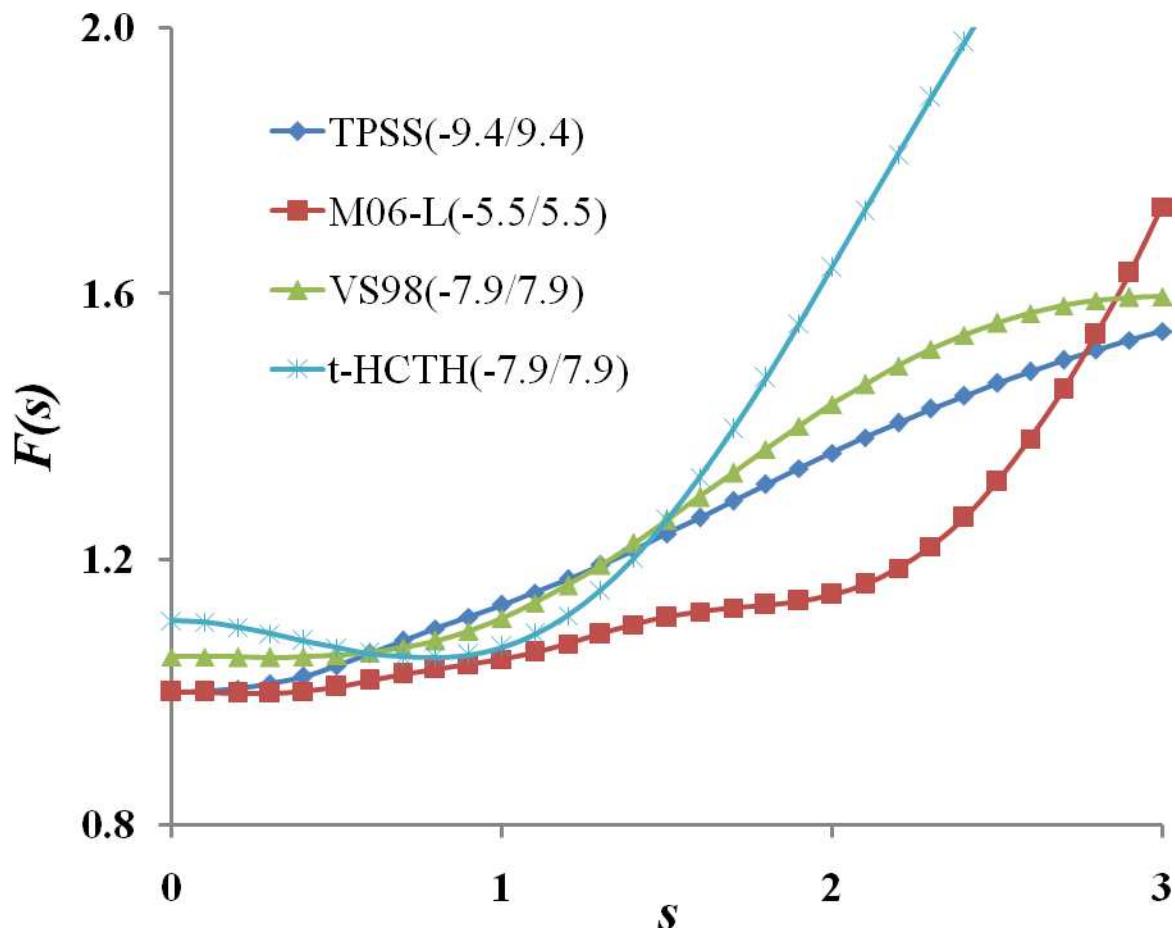


Figure S3. Exchange enhancements factors  $F_X$  as function of reduced gradient  $s$  ( $\alpha = 1.0$ ). (the numbers in the parentheses are the MSEs and MUEs of meta-GGA functionals for the ABDE4 database in units of kcal/mol)

Exchange enhancement factors  $F_X$  of popular GGAs:

$$F_X^{B88} = 1 + \frac{1.07466 \times 0.25237 s^2}{1 + 0.196448 s \sinh^{-1} 7.79555 s}$$

$$F_X^{PW91} = 1 + \frac{1.07466 \times \left( 0.25237 s^2 - 0.14038024 s^2 e^{-100s^2} - 0.00369307 s^4 \right)}{1 + 0.196448 s \sinh^{-1} 7.79555 s - 0.00369688 s^4}$$

$$F_X^{PBE} = 1 + 0.804 - \frac{0.804}{1 + 0.21951 s^2 / 0.804}$$

$$F_X^{revPBE} = 1 + 1.245 - \frac{1.245}{1 + 0.21951 s^2 / 1.245}$$

$$F_X^{RPBE} = 1 + 0.804 \left( 1 - e^{-0.21951 s^2 / 0.804} \right)$$

$$F_X^{PBESol} = 1 + 0.804 - \frac{0.804}{1 + 0.12346 s^2 / 0.804}$$

$$F_X^{SOGGA} = 1 + 0.552 \left( 1 - \frac{0.5}{1 + 0.12346 s^2 / 0.552} - 0.5 e^{-0.12346 s^2 / 0.552} \right)$$

$$F_X^{WC06} = 1 + 0.804 - \frac{0.804}{1 + 0.21951 x / 0.804}$$

$$x = \frac{10}{81} s^2 + \left( 0.21951 - \frac{10}{81} \right) s^2 e^{-s^2} + \ln(1 + 0.0079325 s^2)$$

Exchange enhancement factors  $F_X$  of 4 Meta-GGAs:

TPSS:

$$F_X^{TPSS} = 1 + 0.804 - \frac{0.804}{1 + 0.21951x/0.804}$$

$$x = \left\{ \left[ \frac{10}{81} + 1.5906 \frac{z^2}{(1+z^2)^2} \right] s^2 + \frac{146}{2025} q_b^2 - \frac{73}{405} q_b \sqrt{\frac{1}{2} \left( \frac{3}{5} z \right)^2 + \frac{1}{2} s^4} + \frac{1}{0.804} \left( \frac{10}{81} \right)^2 s^4 + 2\sqrt{1.537} \frac{10}{81} \frac{3}{5} z^2 + 1.537 \times 0.21951 s^6 \right\} / (1 + \sqrt{1.537} s^2)^2$$

$$q_b = \frac{9}{20} \frac{\alpha - 1}{(1 + 0.40\alpha(\alpha - 1))^{1/2}} + \frac{2}{3} s^2$$

$$z = \frac{5s^2}{3\alpha + 5s^2}$$

M06-L:

$$F_X^{M06-L} = F_X^{PBE} f(\omega) + h(x, z)$$

$$\omega = 1 - \frac{2\alpha + 10s^2/3}{1 + \alpha + 5s^2/3}$$

$$x = 7.79555s$$

$$z = \frac{3}{5} (6\pi^2)^{2/3} (\alpha + 5s^2/3 - 1)$$

$$\gamma(x, z) = 1 + 0.001867(x^2 + z)$$

The expressions of  $f(\omega)$  and  $h(x, z)$  are in the M06-L paper.

See: Y. Zhao and D. G. Truhlar, J. Chem. Phys. **125**, 194101 (2006).

VS98:

$$F_X^{VS98} = -1.07466 \left( \frac{-0.9800}{\gamma(x, z)} + \frac{-0.003557x^2 + 0.006250z}{\gamma^2(x, z)} + \frac{-0.00002354x^2 - 0.0001283x^2z + 0.0003575z^2}{\gamma^3(x, z)} \right)$$

$$x = 7.79555s$$

$$z = \frac{3}{5} (6\pi^2)^{2/3} (\alpha + 5s^2/3 - 1)$$

$$\gamma(x, z) = 1 + 0.001867(x^2 + z)$$

$\tau$ -HCTH:

$$F_X^{\tau-HCTH} = g_{local} + g_{non-local} f(\omega)$$

$$u(s) = \frac{0.004 \times (7.795554s)^2}{1 + 0.004 \times (7.795554s)^2}$$

$$\omega = 1 - \frac{2\alpha + 10s^2/3}{1 + \alpha + 5s^2/3}$$

The expressions of  $g_{local}$  and  $g_{non-local}$  is in the  $\tau$ -HCTH paper.

See: A. D. Boese and N. C. Handy, J. Chem. Phys. **116**, 9559 (2002).